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α-Ketoglutarate Dehydrogenase (α-KGDH) Activity Assay Kit

Note: Take two or three different samples for prediction before test.

Detection equipment: Spectrophotometer/Microplate reader

Cat No: AK0281 **Size:** 100T/96S

Components

Reagent I: 110 mL×1, store at 4°C;

Reagent II: 1 mL×1, store at -20°C;

Reagent III: 22 mL×1, store at 4°C;

Reagent IV: Powder×1, store at 4°C;

Reagent V: Powder×1, store at 4°C;

Reagent VI: Powder×1, store at -20°C;

Reagent VII: Powder×1, store at -20°C;

Reagent VIII: Powder×1, store at -20°C and protect from light. Add 0.8 mL of distilled water when the solution will be used, the unused reagents are stored at -20°C.

Preparation of working solution: when the solution will be used, transfer Reagent IV, V, VI and VII to Reagent III, mix and dissolve them for use.

Description

 α -Ketoglutarate Dehydrogenase (α -KGDH, EC 1.2.4.2) is one of the key enzymes in the regulation of tricarboxylic acid cycle and widely exists in mitochondria of animal, plant, microorganisms and cultured cells, which catalyzes the oxidative decarboxylation of α -ketoglutarate to succinyl coenzyme A.

 α -KGDH catalyzes α -ketoglutarate, NAD⁺ and coenzyme A to form succinyl coenzyme A, carbon dioxide and NADH. NADH has a characteristic absorption peak at 340 nm. The activity of α -KGDH is expressed by the formation rate of NADH.

Required but not provided

Ultraviolet spectrophotometer/microplate reader, water-bath, tabletop centrifuge, adjustable pipette, micro quartz cuvette/96 well flat-bottom plate (UV plate), mortar/homogenizer, ice and distilled water.

Protocol

I. Extraction of α-KGDH:

Accurately weigh 0.1 g of tissue or collect 5 million cells, add 1 mL of Reagent I and 10 μ L of Reagent II, homogenize by using homogenizer/mortar in ice bath, fully grind, centrifuge at 11000 \times g for 10 minutes at 4°C, take the supernatant, place it on ice for test.

II. Procedure

1. Preheat Spectrophotometer/Microplate reader for 30 minutes, adjust wavelength to 340 nm, set zero

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with distilled water.

2. Blank tube:

Take 200 µL of working solution and add it to the micro quartz cuvette or 96 well flat-bottom plate, incubate it at 37°C for 5 minutes, then take out the cuvette, add 8 µL of Reagent VIII and 12 µL of distilled water in turn into the cuvette, mix them well and immediately measure the absorbance value A1 of 0 s at 340 nm, react accurately at 37°C for 2 minutes, record the absorbance value A2 of 2 minutes at 340 nm, calculate $\Delta A_B = A2-A1$.

3. Measuring tube:

Take 200 µL of working solution and add it to the micro quartz cuvette or 96 well flat-bottom plate, incubate it at 37°C (mammal) or 25°C (other species) for 5 minutes, then take out the cuvette, add 8 µL of Reagent VIII and 12 µL of samples in turn into the cuvette, mix them well and immediately measure the absorbance value A3 of 0s at 340 nm, react accurately 37°C (mammal) or 25°C (other species) for 2 minutes, and record the absorbance value A4 of 2 minutes at 340 nm, Calculate $\Delta A_T = A4-A3$.

III. Calculation of α-KGDH activity

A. The calculation formula according to the determination of micro quartz cuvette

(1) Protein concentration

Unit definition: One unit of enzyme activity is defined as the amount of enzyme catalyzes the production of 1 nmol of NADH per minute in the reaction system every milligram tissue protein.

$$\alpha\text{-KGDH(U/mg prot)} \!=\! [(\Delta A_T \!-\! \Delta A_B) \div (\epsilon \times d) \times V_{RV} \times 10^9] \div (Cpr \times V_{SV}) \div T = 1473.7 \times (\Delta A_T \!-\! \Delta A_B) \div Cpr \ (\Delta A_T + \Delta A_T + \Delta$$

2) Sample weight

Unit definition: One unit of enzyme activity is defined as the amount of enzyme catalyzes the production of 1 nmol of NADH per minute in the reaction system every gram tissue.

$$\alpha\text{-KGDH}\left(\text{U/g fresh weight}\right) = \left[\left(\Delta A_T - \Delta A_B\right) \div \left(\epsilon \times d\right) \times V_{RV} \times 10^9\right] \div \left(V_{SV} \div V_{STV} \times W\right) \div T = 1488.5 \times \left(\Delta A_T - \Delta A_B\right) \div W$$

(3) Germ or cells

Unit definition: One unit of enzyme activity is defined as the amount of enzyme catalyzes the production of 1 nmol of NADH per minute in the reaction system every 10 thousand germ or cells.

$$\alpha\text{-KGDH }(\text{U}/10^4\,\text{cell}) = [(\Delta A_T - \Delta A_B) \div (\epsilon \times d) \times V_{RV} \times 10^9] \div (V_{SV} \div V_{STV} \times 500) \div T = 2.977 \times (\Delta A_T - \Delta A_B)$$

 V_{RV} : The total volume of reaction system, $2.2 \times 10^{-4} L$;

ε: The molar extinction coefficient of NADH, 6.22×10³ L/mol/cm;

d: cuvette light diameter, 1 cm;

V_{SV}: sample volume, 0.012 mL;

V_{STV}: The volume of Reagent I and Reagent II, 1.01 mL;

T: reaction time, 2 minutes;

Cpr: The concentration of sample protein, mg/mL;

W: Sample weight, g.

500: Cells or germ, 5 million.







- B. The calculation formula according to the determination of 96 well plate:
- (1) Protein concentration

Unit definition: One unit of enzyme activity is defined as the amount of enzyme catalyzes the production of 1 nmol of NADH per minute in the reaction system every milligram tissue protein.

$$\alpha\text{-KGDH(U/mg prot)} = [(\Delta A_T - \Delta A_B) \div (\epsilon \times d) \times V_{RV} \times 10^9] \div (Cpr \times V_{SV}) \div T = 2456.2 \times (\Delta A_T - \Delta A_B) \div Cpr$$

(2) Sample weight

Unit definition: One unit of enzyme activity is defined as the amount of enzyme catalyzes the production of 1 nmol of NADH per minute in the reaction system every gram tissue.

$$\alpha\text{-KGDH} \quad (\text{U/g fresh weight}) = [(\Delta A_T - \Delta A_B) \div (\epsilon \times d) \times V_{RV} \times 10^9] \div (V_{SV} \div V_{STV} \times W) \div T = 2480.7 \times (\Delta A_T - \Delta A_B) \div W$$

(3) Germ or cells

Unit definition: One unit of enzyme activity is defined as the amount of enzyme catalyzes the production of 1 nmol of NADH per minute in the reaction system every 10 thousand germ or cells.

$$\alpha\text{-KGDH} \quad (\text{U}/10^4 \text{ cell}) = [(\Delta A_\text{T} - \Delta A_\text{B}) \div (\epsilon \times d) \times V_{\text{RV}} \times 10^9] \div (V_{\text{SV}} \div V_{\text{STV}} \times 500) \div T = 4.962 \times (\Delta A_\text{T} - \Delta A_\text{B})$$

 V_{RV} : total volume of reaction system, 2.2×10-4L;

ε: The molar extinction coefficient of NADH, 6.22×10³ L/mol/cm;

d: cuvette light diameter, 0.6 cm;

V_{SV}: sample volume, 0.012 mL;

V_{STV}: The volume of Reagent I and Reagent II, 1.01 mL;

T: reaction time, 2 minutes;

Cpr: The concentration of sample protein, mg/mL;

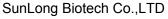
W: Sample weight, g.

500: Cells or germ, 5 million.

Note:

- 1. All reagents and samples shall be placed on ice during the determination to avoid denaturation and deactivation.
- 2. The temperature of the reaction solution in the cuvette must be kept at 37°C or 25°C. Take a small beaker and put it into a certain amount of 37°C or 25°C distilled water. Put the beaker into a 37°C or 25°C water bath. Put the cuvette and reaction solution into the beaker during the reaction.
- 3. It is better for two people to do the experiment at the same time, one for color comparison and one for timing, so as to ensure the accuracy of the experimental results.
- 4. The ΔA value of the test tube is between 0.01-0.25. If the ΔA value of the test tube is greater than 0.25, the sample shall be diluted.
- 5. As the Reagent I contents a certain concentration of protein (about 1 mg/mL), the protein content of the extract solution itself needs to be subtracted when determining the protein concentration of the sample.

Experimental example:









1. Take 0.1 g of barnyardgrass for sample treatment, dilute the supernatant for 2 times, and then operate according to the determination steps. Use micro quartz colorimetric plate to measure and calculate $\Delta A_T = A4-A3 = 0.3243-0.3115 = 0.0128$, $\Delta A_B = A2-A1=0$

 α -KGDH (U/g mass) = 1488.5 × (ΔA_T - ΔA_B) × W × 2 (dilution ratio) = 381.056 U/g mass.

2. After centrifugation at 4°C for 10 min, the supernatant was taken and operated according to the determination steps. The results were as follows: $\Delta A = A4 - A3 = 1.2123 - 0.9623 = 0.2500$, $\Delta AB = A2 - A1 = 0$ α -KGDH (U/g mass) = $1488.5 \times (\Delta A_T - \Delta A_B) \div W = 3721.25$ U/g mass.

Recent product Citations:

[1] Jianyun Yue, Changjian Du, Jing Ji, et al. Inhibition of α -ketoglutarate dehydrogenase activity afects adventitious root growth in poplar via changes in GABA shunt. Planta. July 2018;(IF3.06)

[2] Xiao Li,Qi Zhao,Jianni Qi,et al. lncRNA Ftx promotes aerobic glycolysis and tumor progression through the PPARγ pathway in hepatocellular carcinoma. International Journal of Oncology. May 2018; (IF3.571)

References:

[1] Park L C H, Calingasan N Y, Sheu K F R, et al. Quantitative α-ketoglutarate dehydrogenase activity staining in brain sections and in cultured cells[J]. Analytical biochemistry, 2000, 277(1): 86-93.

Related Products:

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